

# **POLLUTED EMISSION TREATMENTS FROM INCINERATOR GASES**

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# Ecole de Chimie de Rennes



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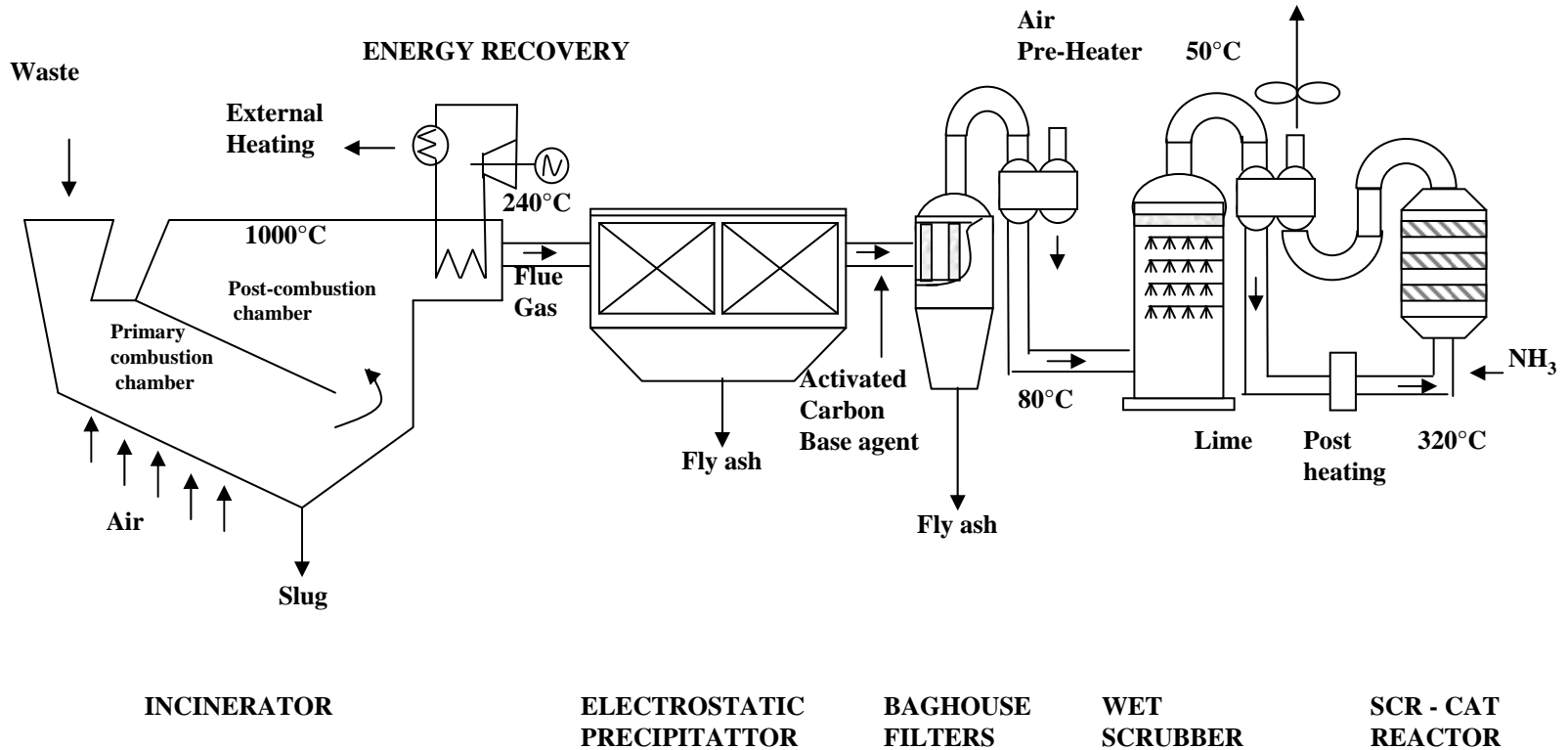


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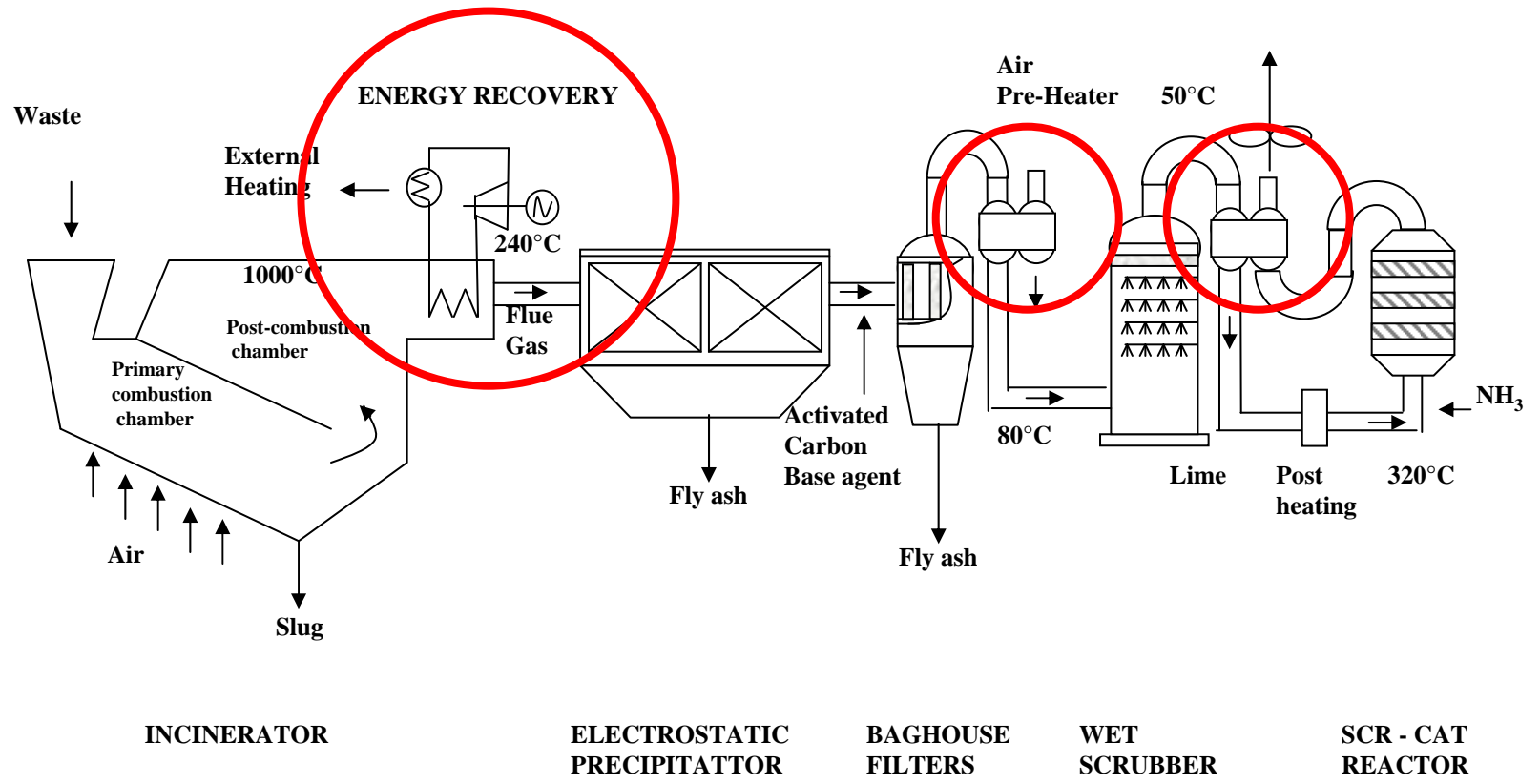
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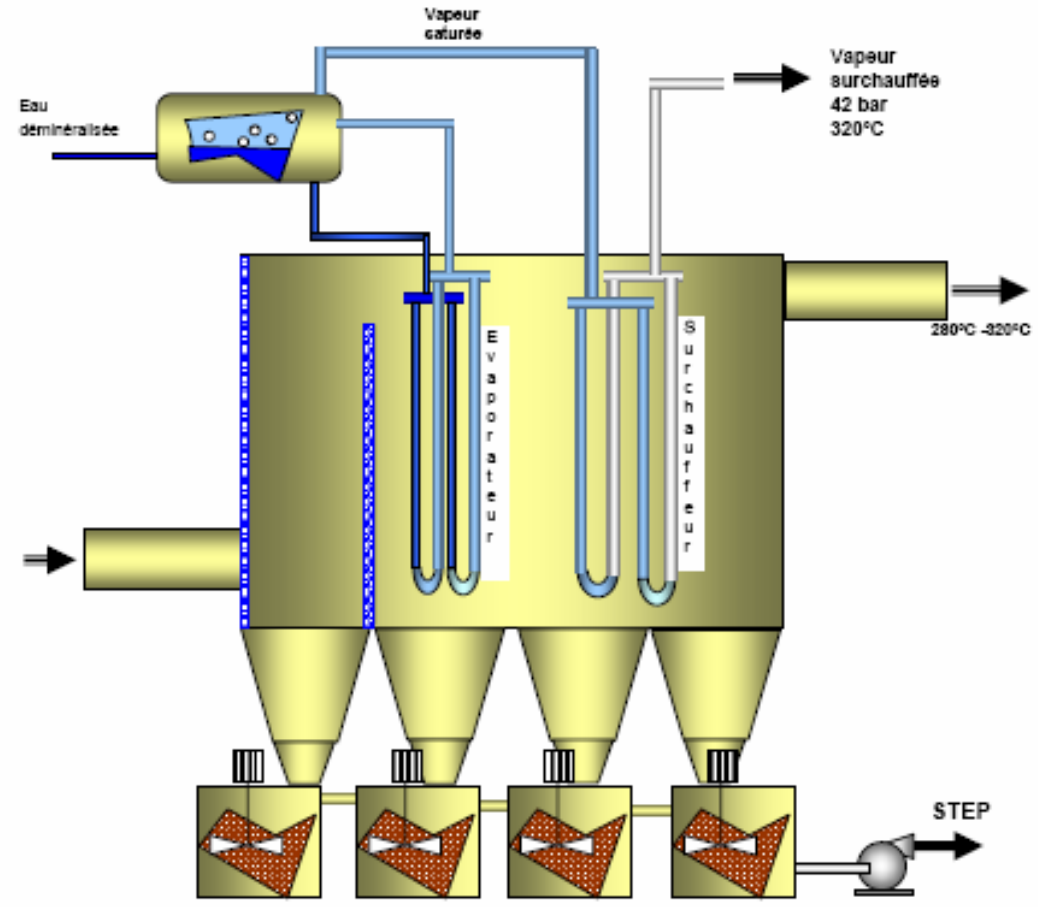


(adapted from Fino *et al.*, 2005)

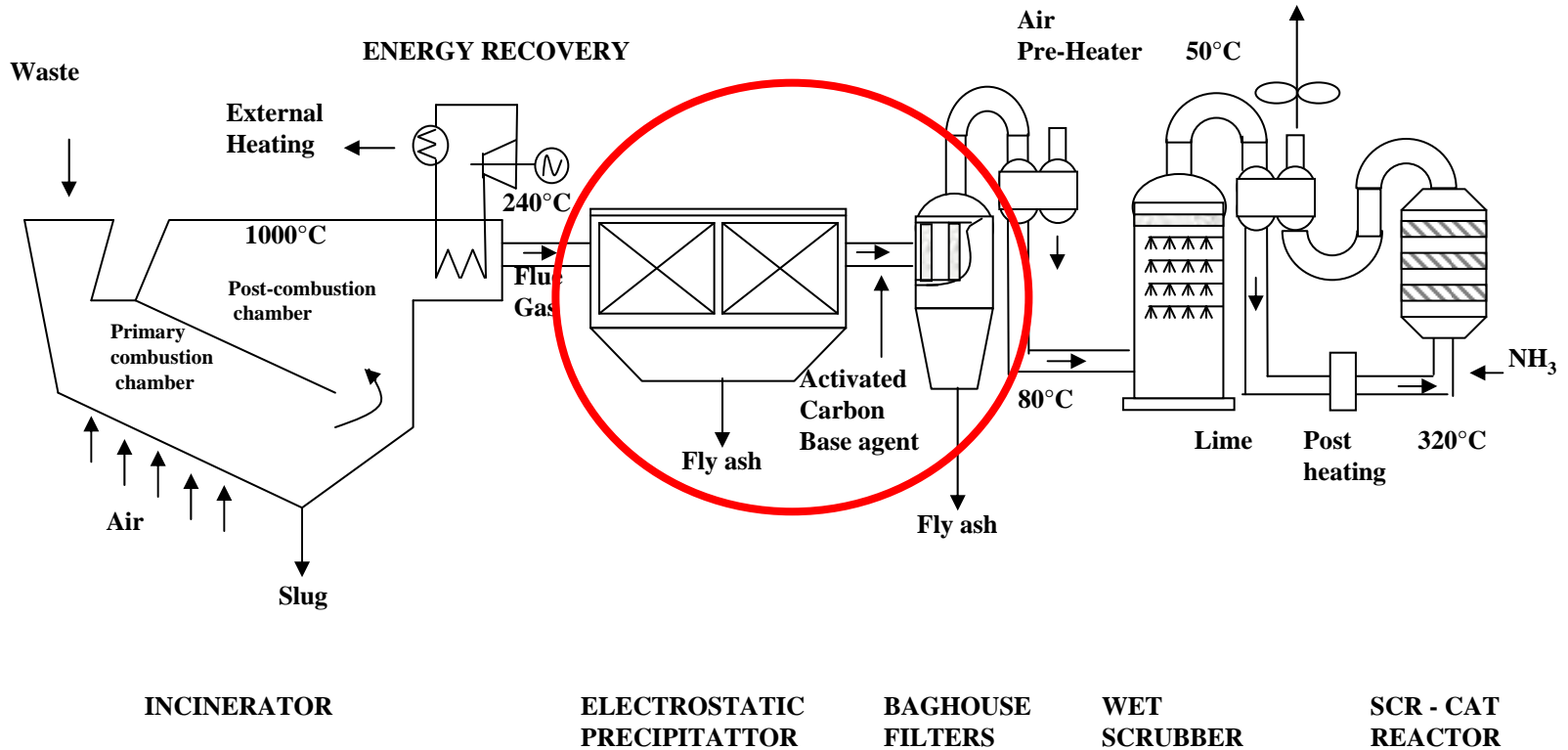


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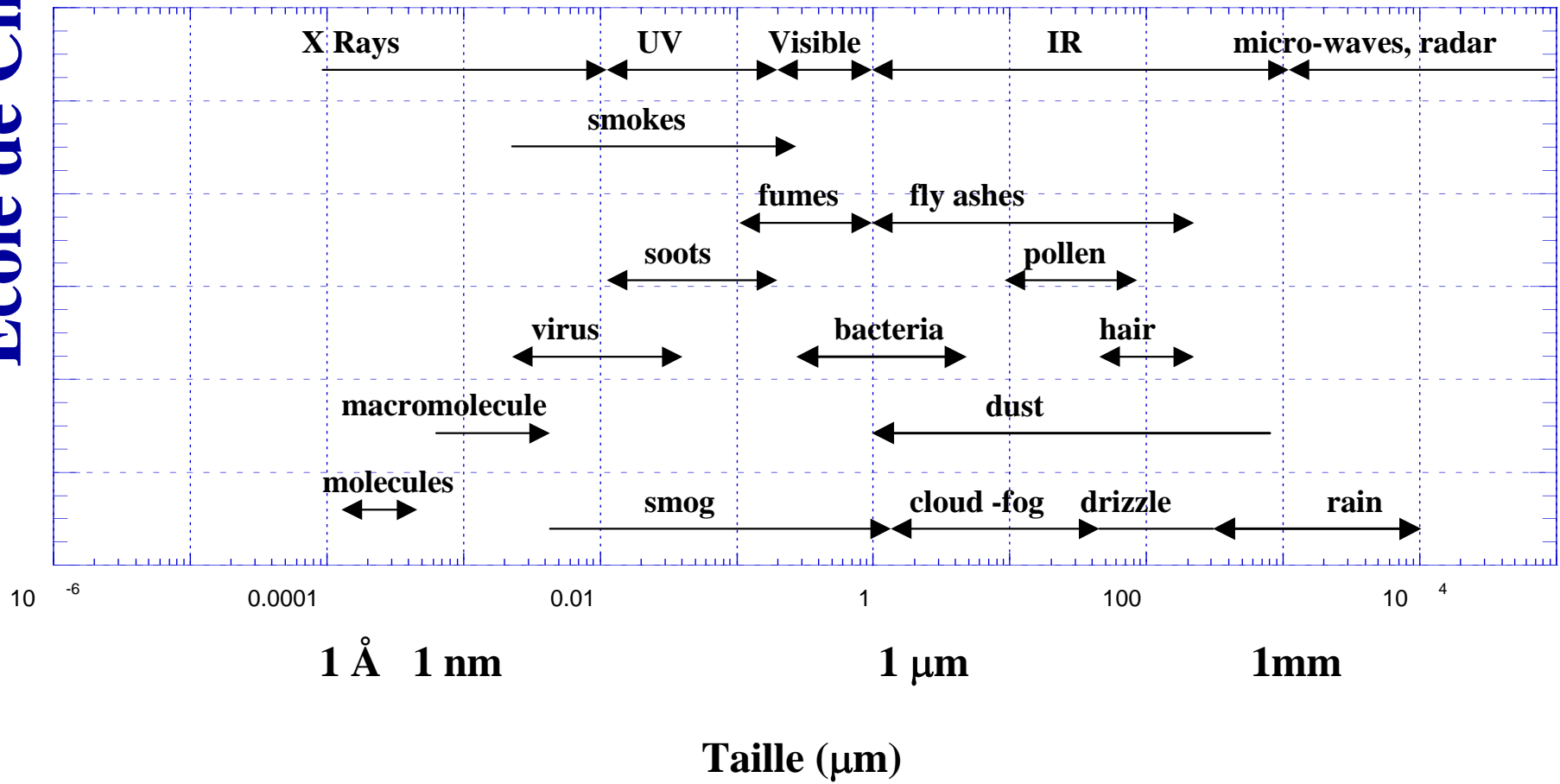
## Energy recovery

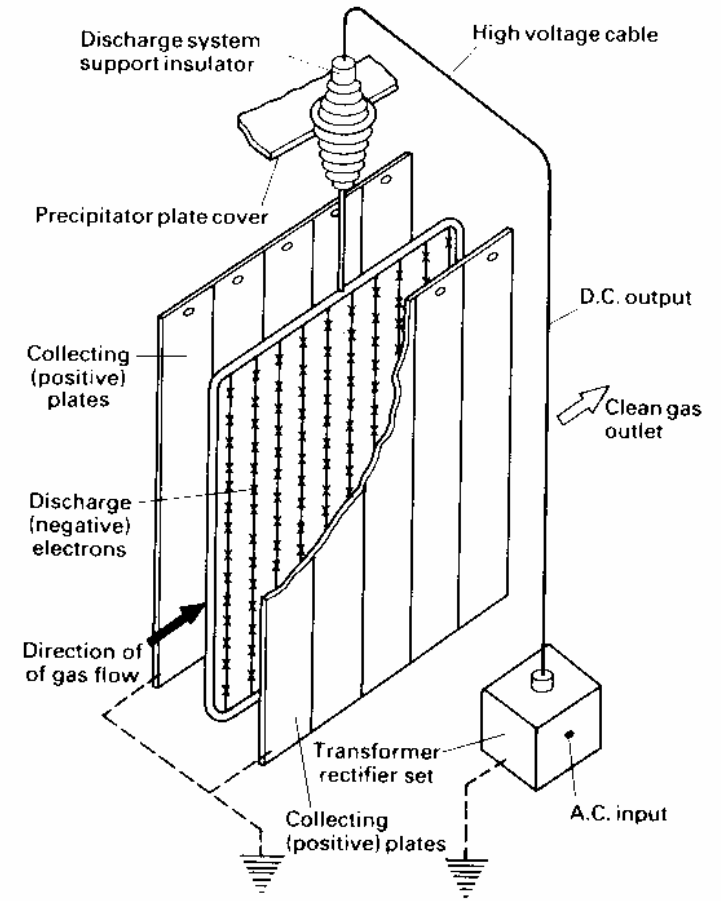
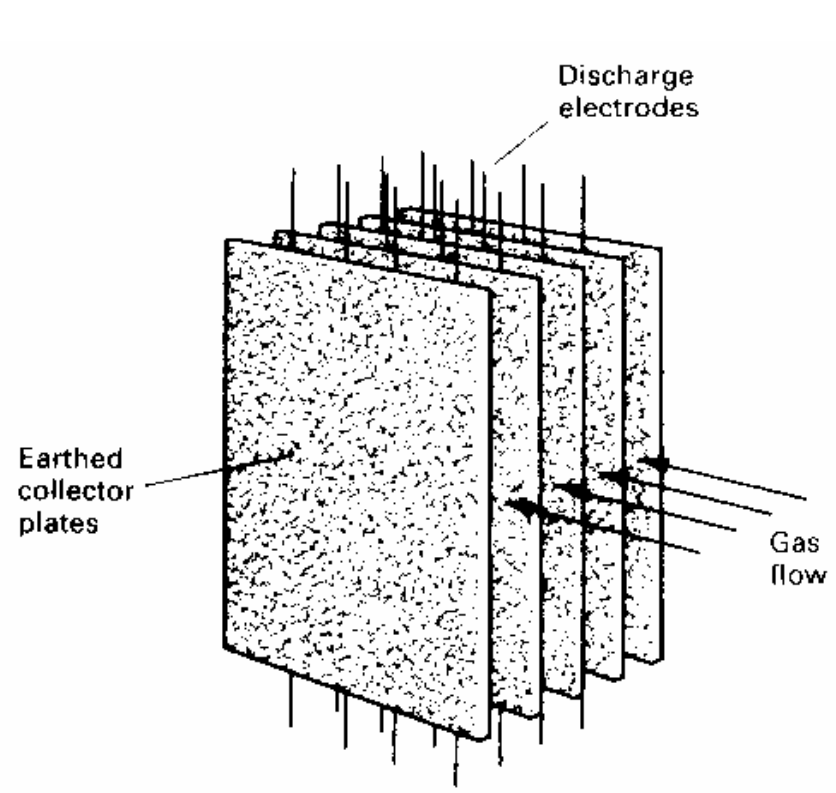




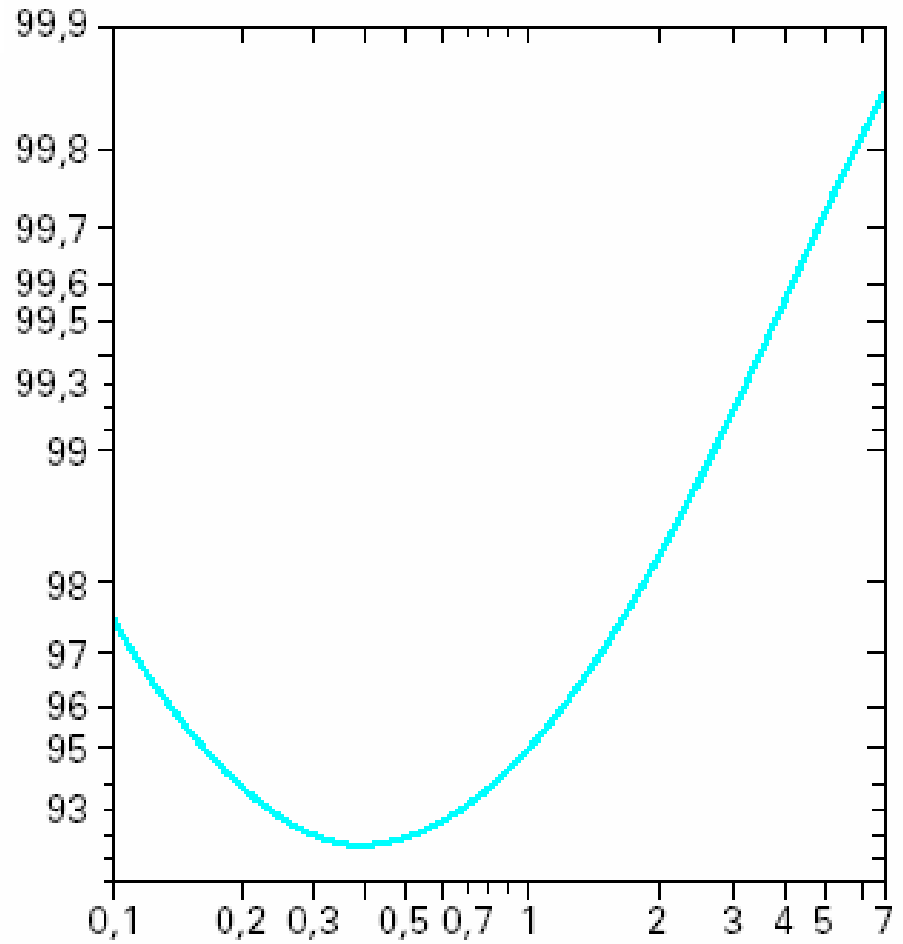


(adapted from Fino *et al.*, 2005)

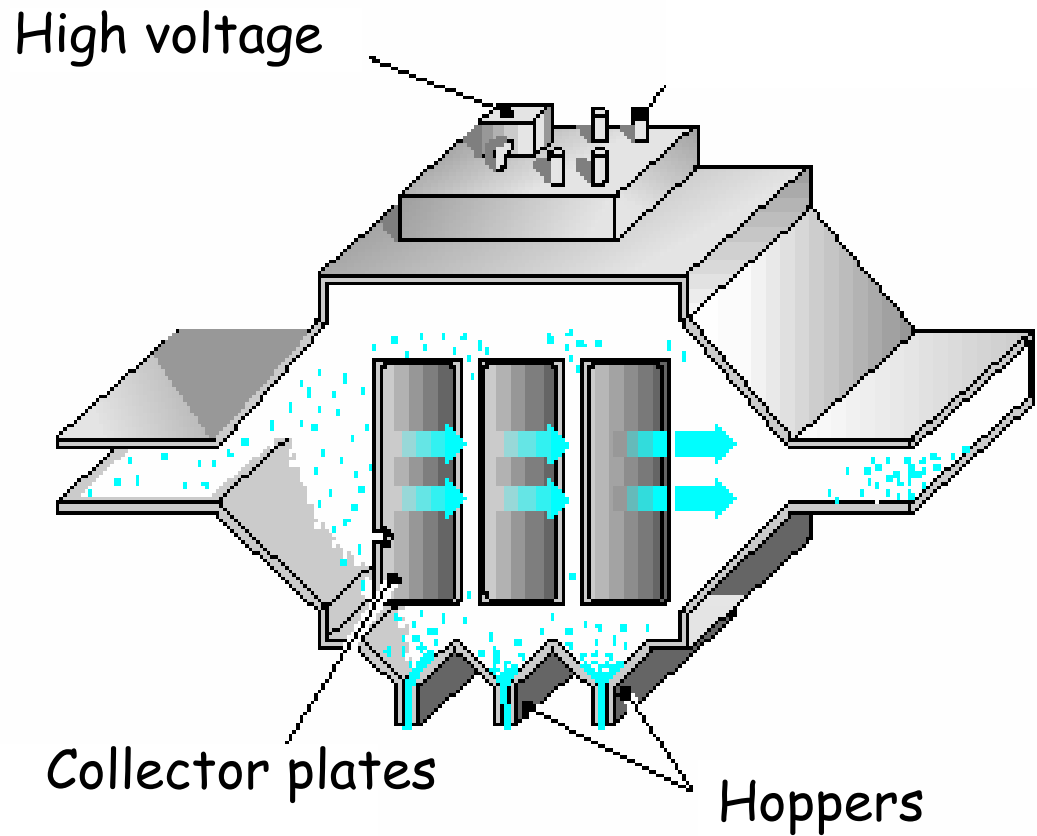




Efficiency (%)



Particle size (µm)



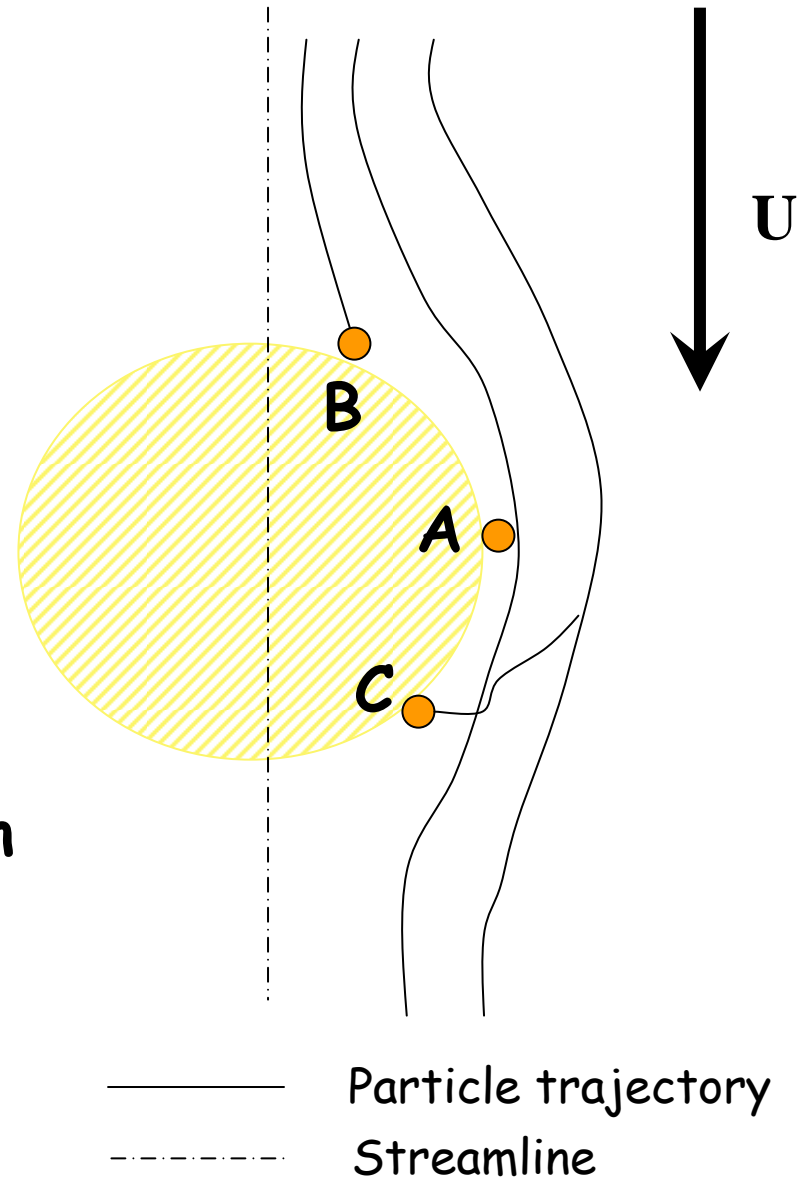


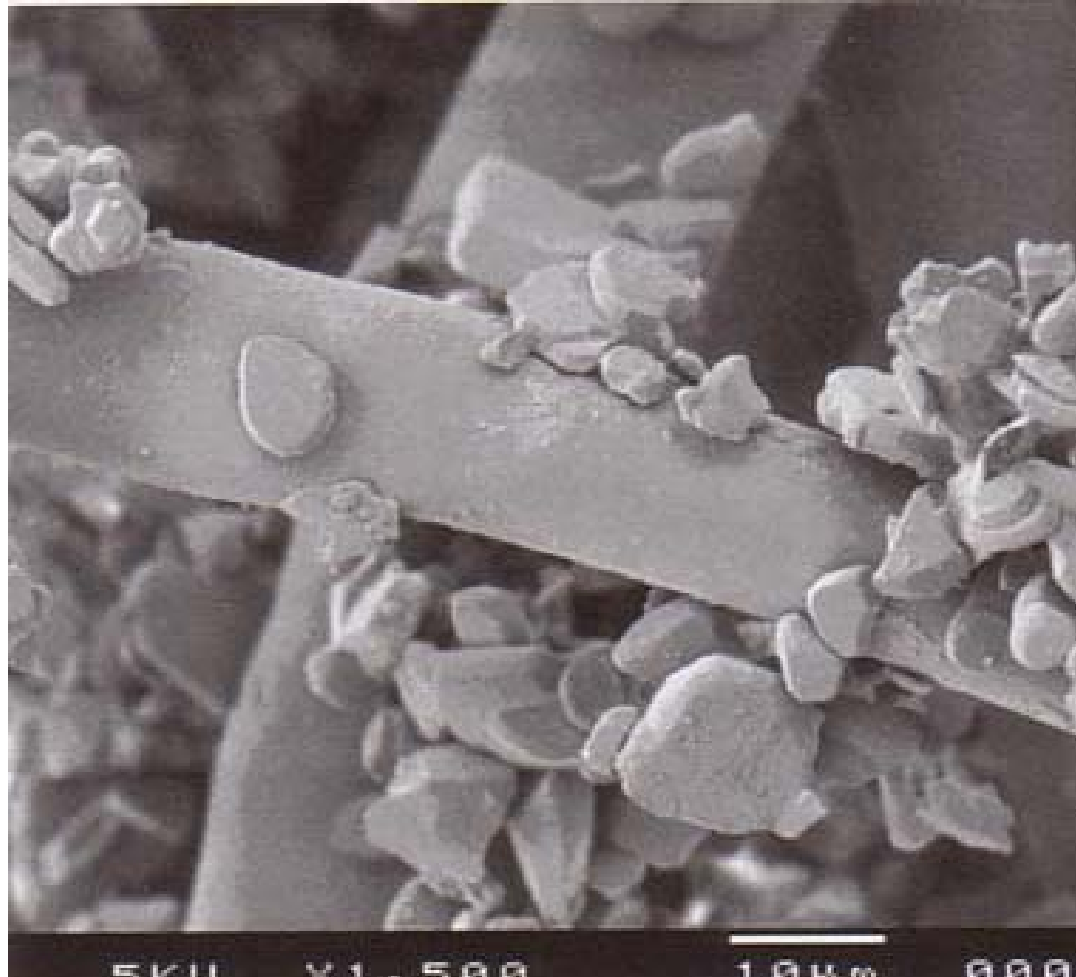


For an isolated single sphere:

**Mode of action of the  
basic mechanisms:**

- A**      **Interception**
- B**      **Sedimentation**
- C**      **Diffusion - Adsorption**

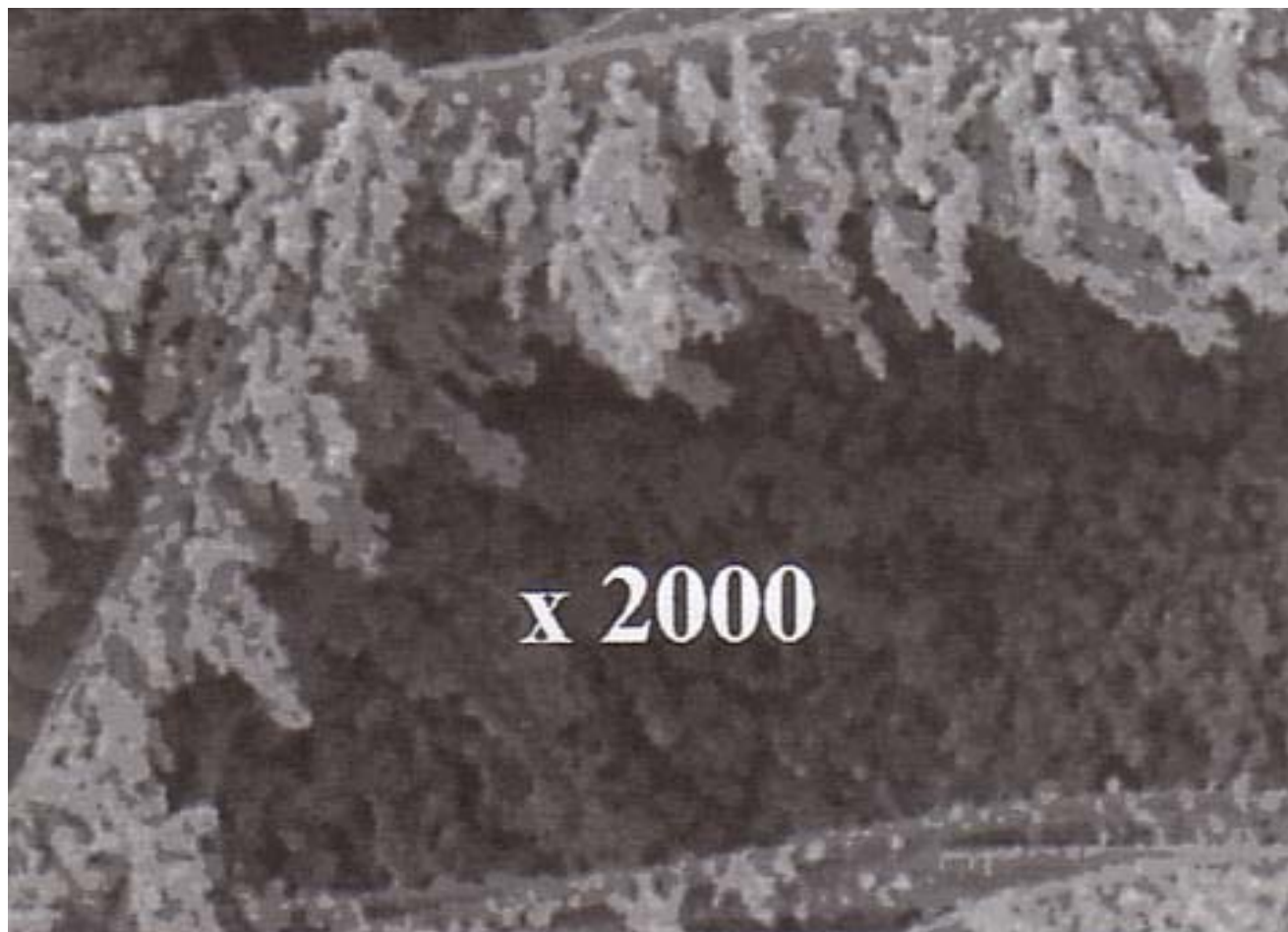




**Particle aggregates on fibers**  
**Particle diameter = 2.6  $\mu\text{m}$**

D. Thomas, 2005

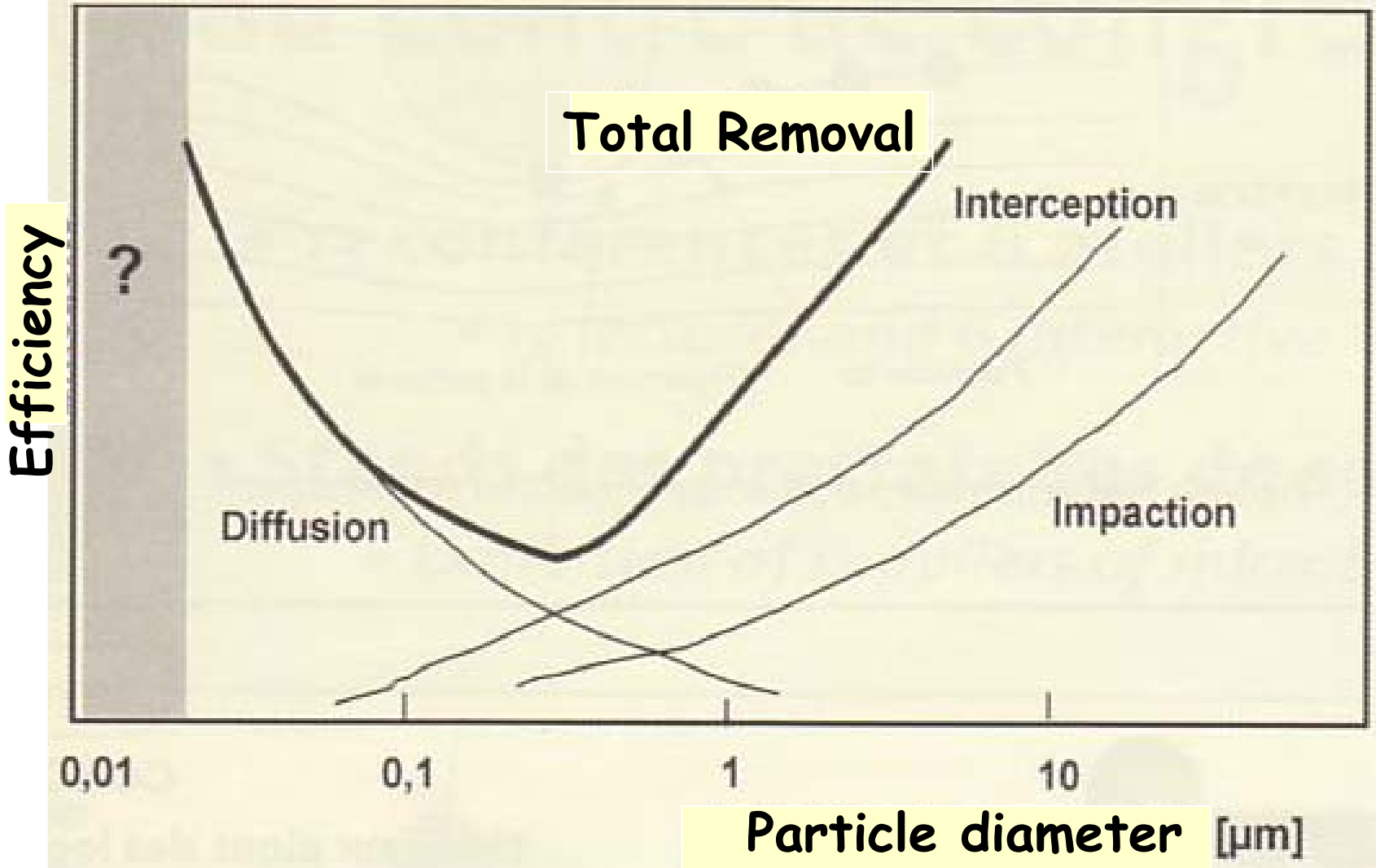
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**Particle dendrites**  
**Particle diameter = 0.15  $\mu\text{m}$**

D. Thomas, 2005

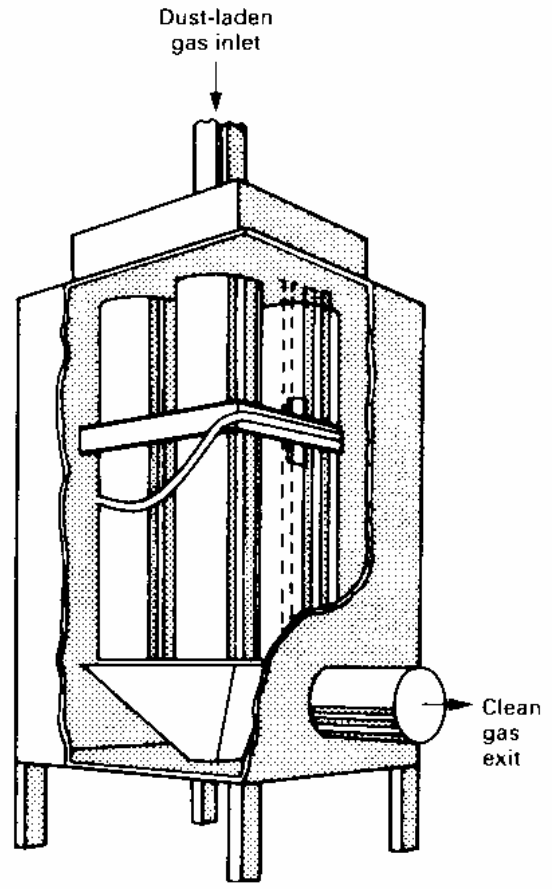
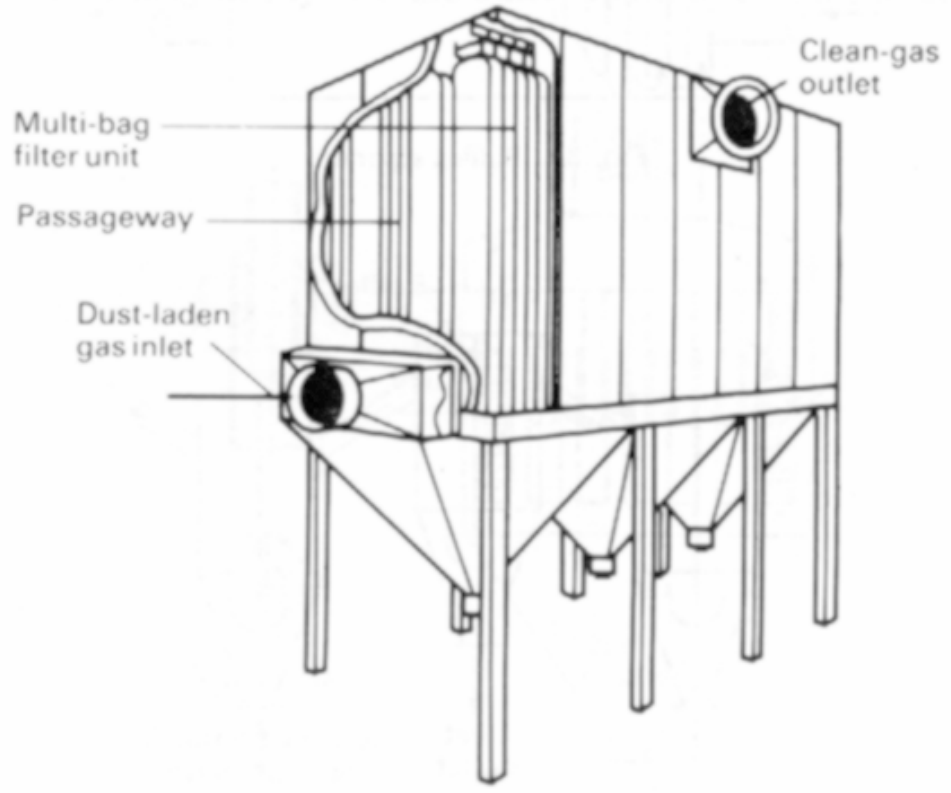
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D. Thomas, 2005

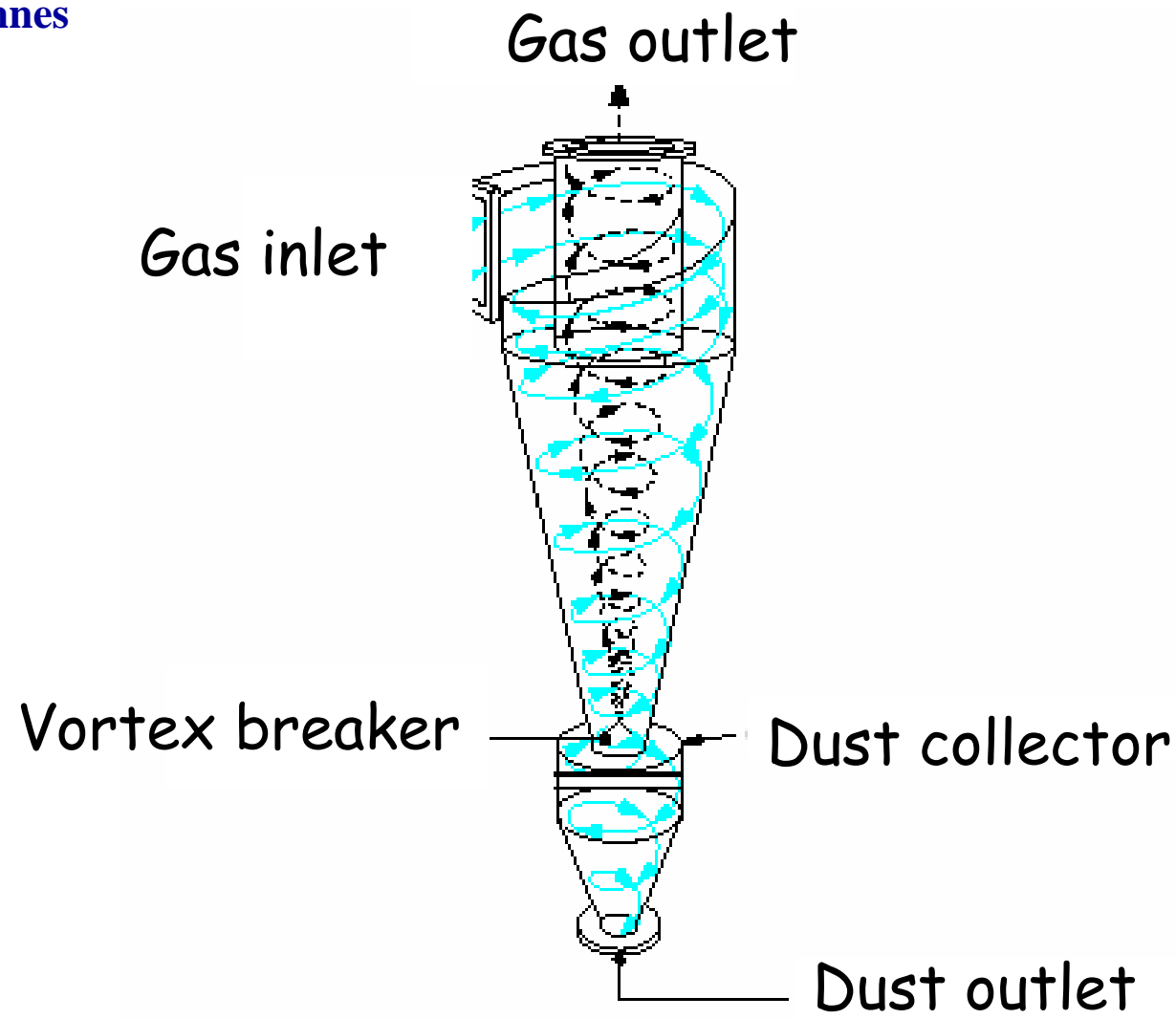
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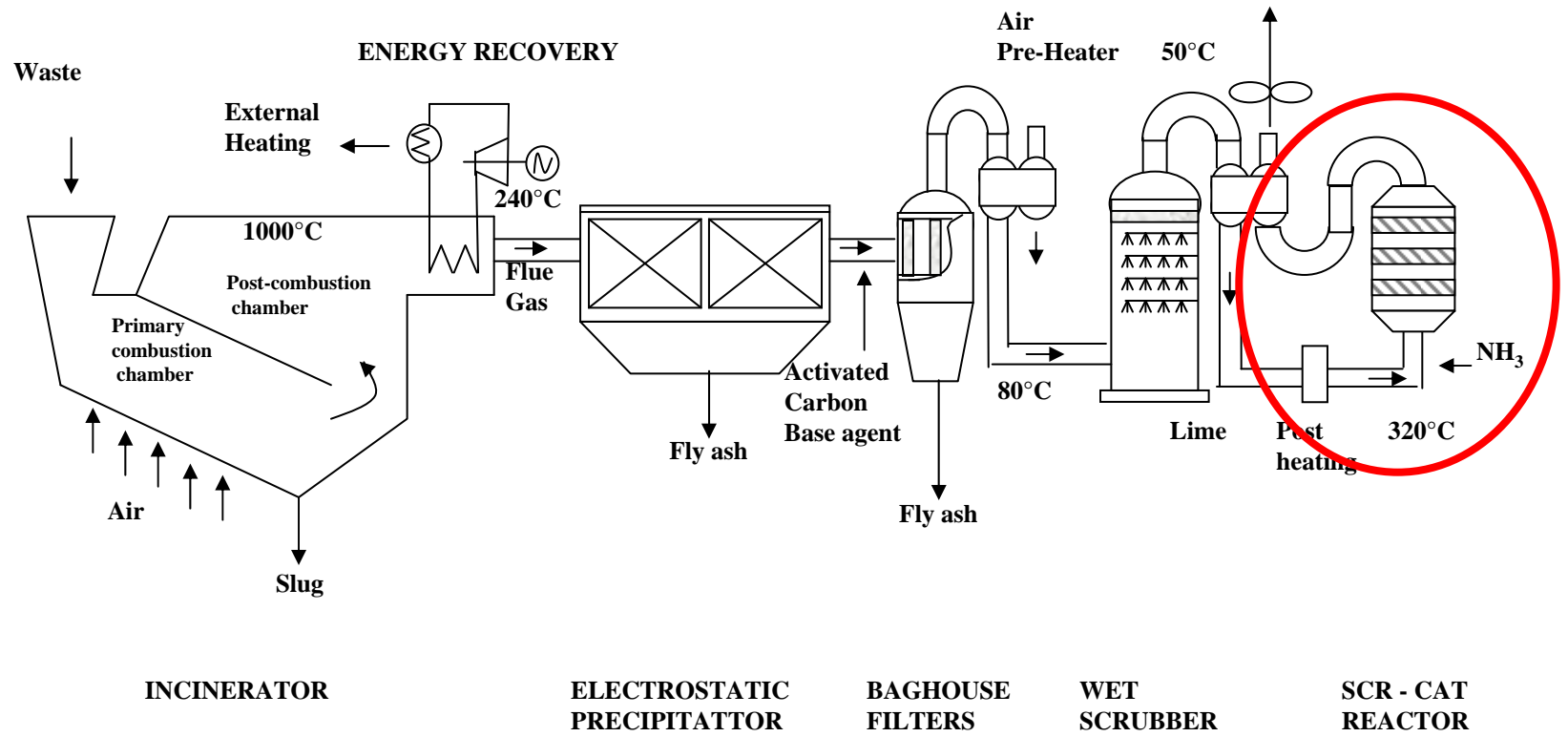
## Industrial system example





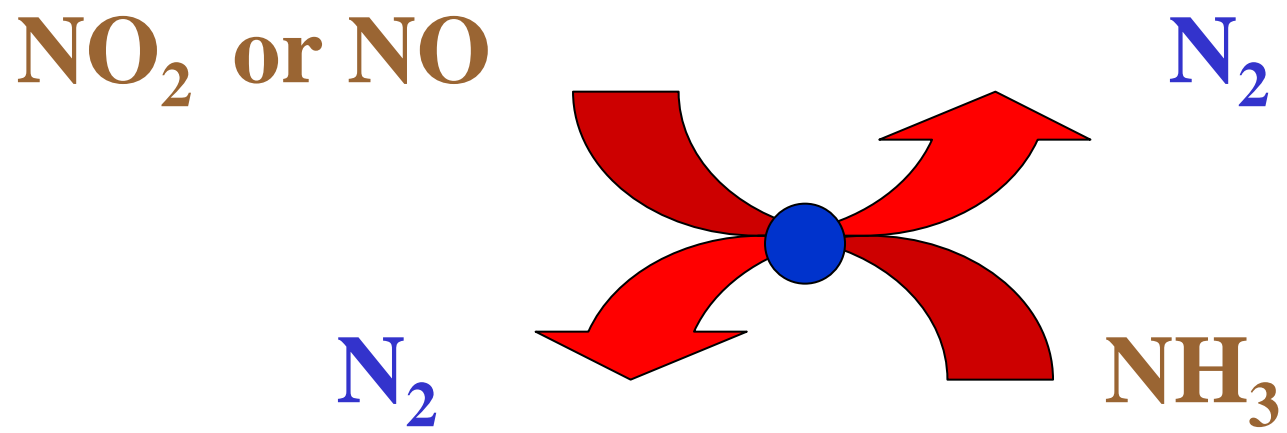


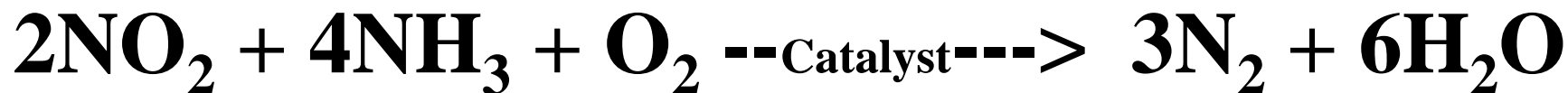
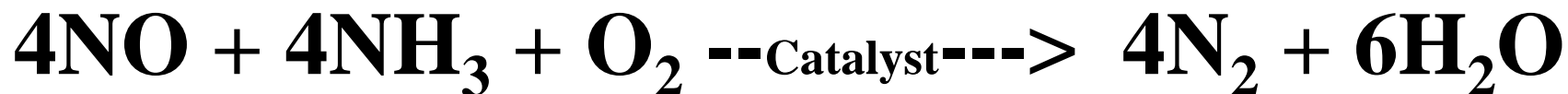
# DeNox



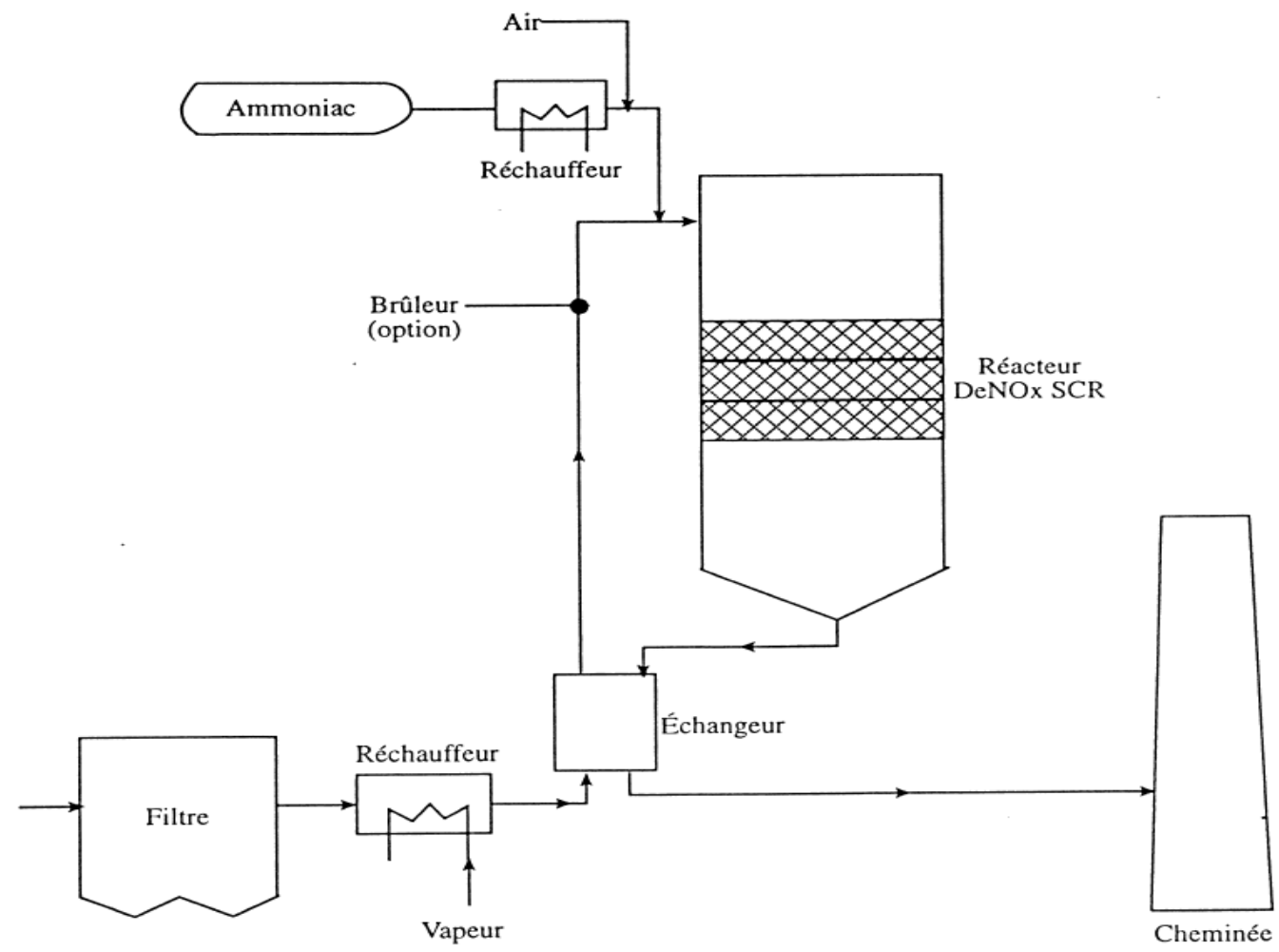
(adapted from Fino *et al.*, 2005)

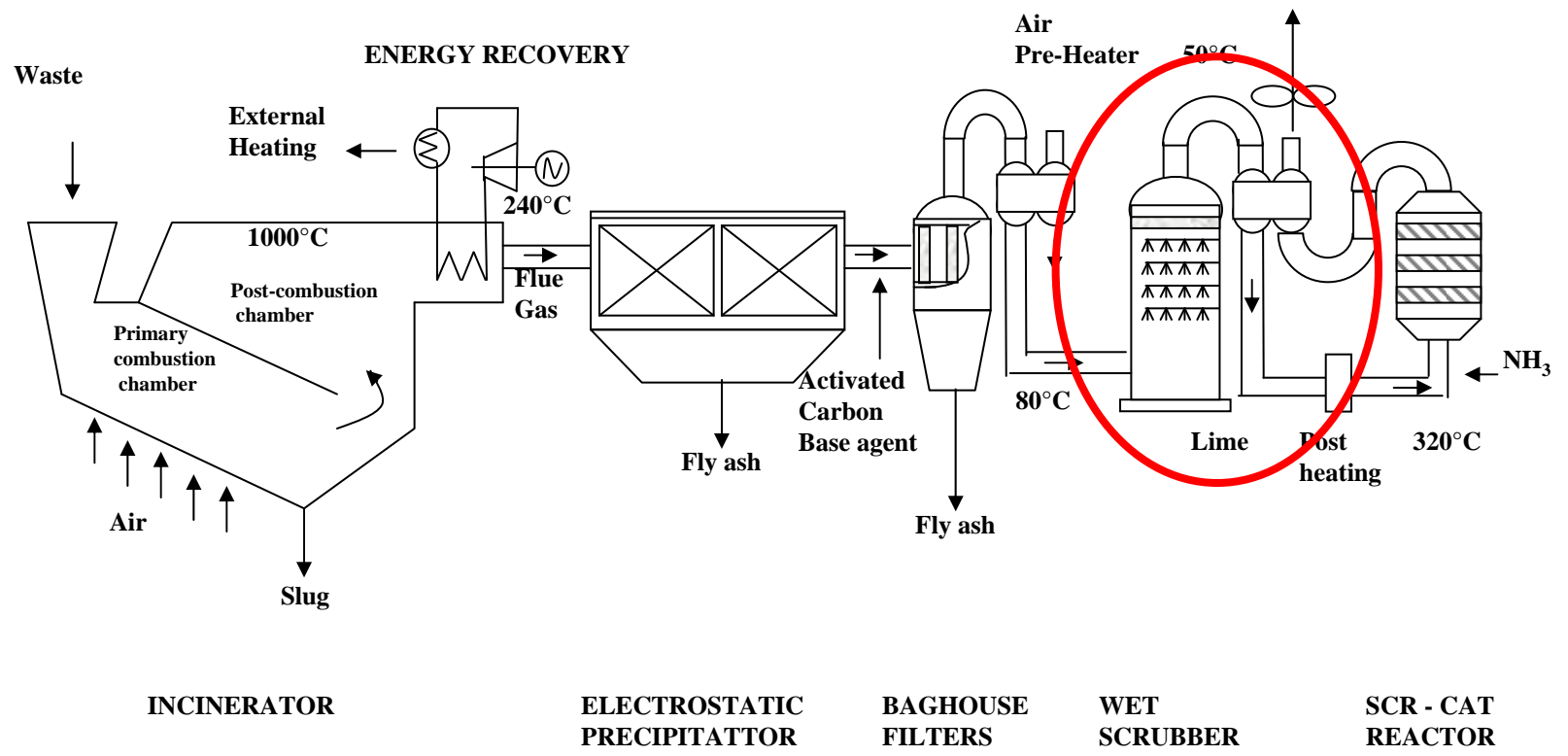






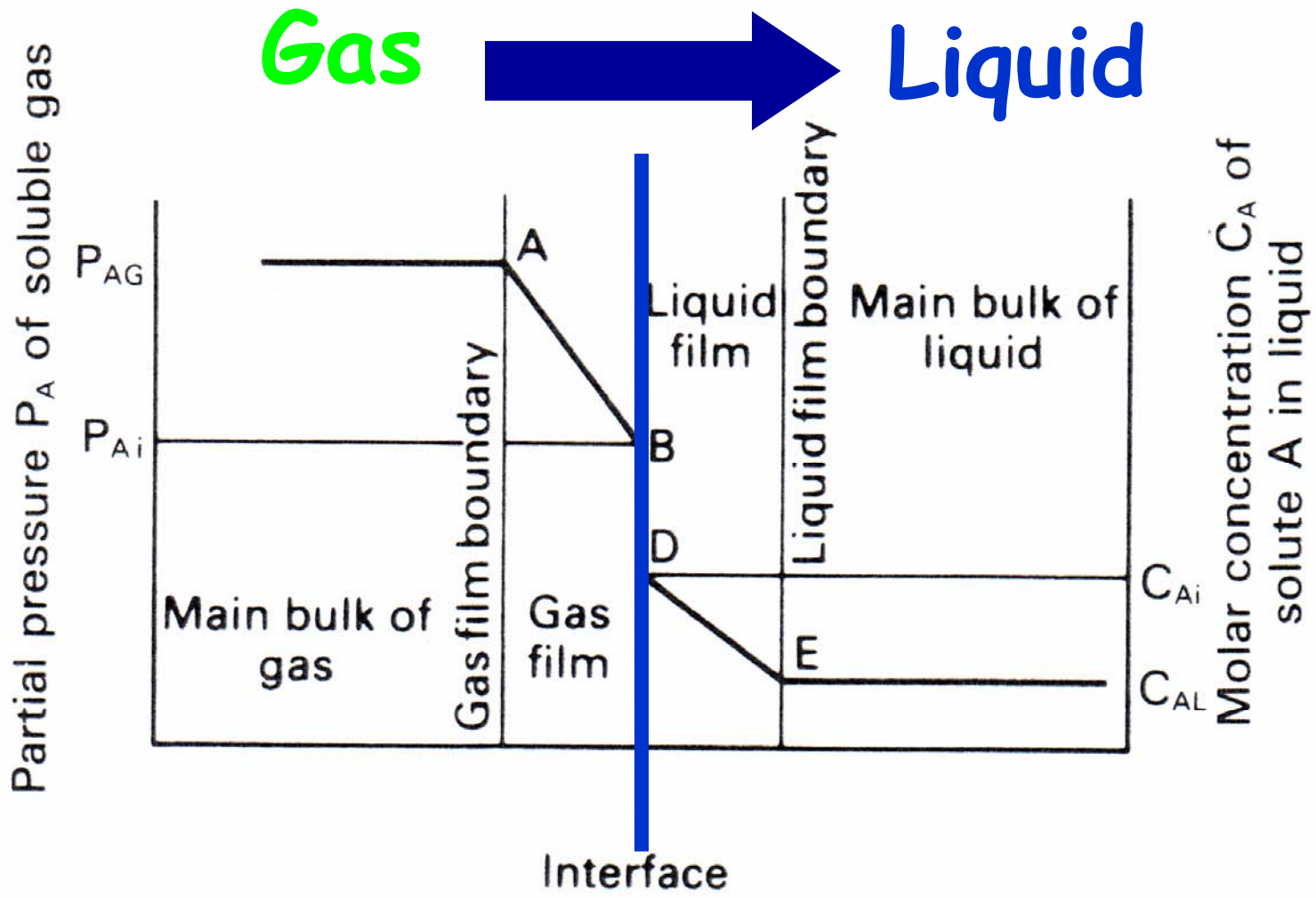
- **Inert support**
  - Alumino silicate
  - Zeolites
- **Metal**
  - Pt
  - Pd, V, Zr...





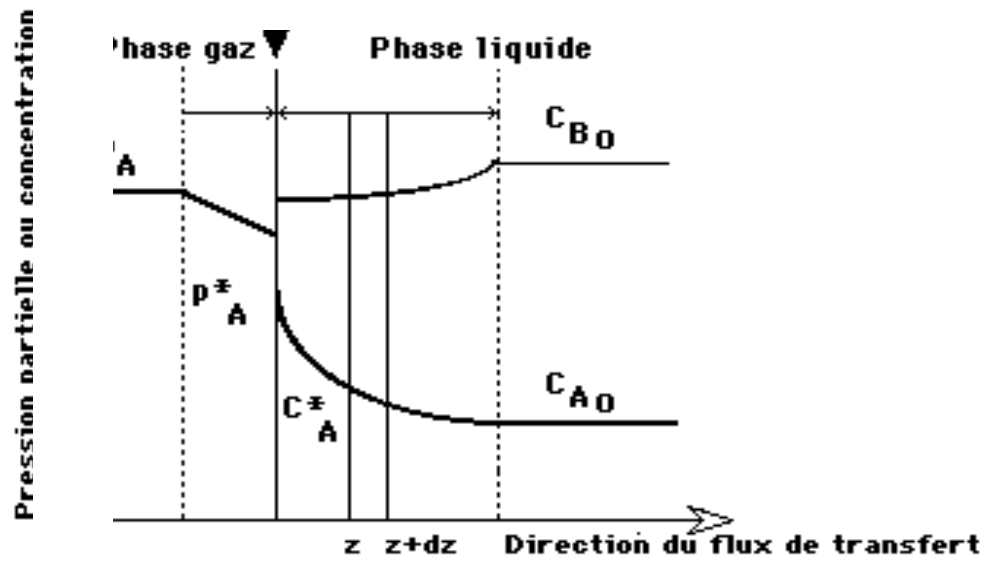
(adapted from Fino *et al.*, 2005)

# Gas - Liquid transfer



Gas  $\longrightarrow$  Liquid

G-L Interface

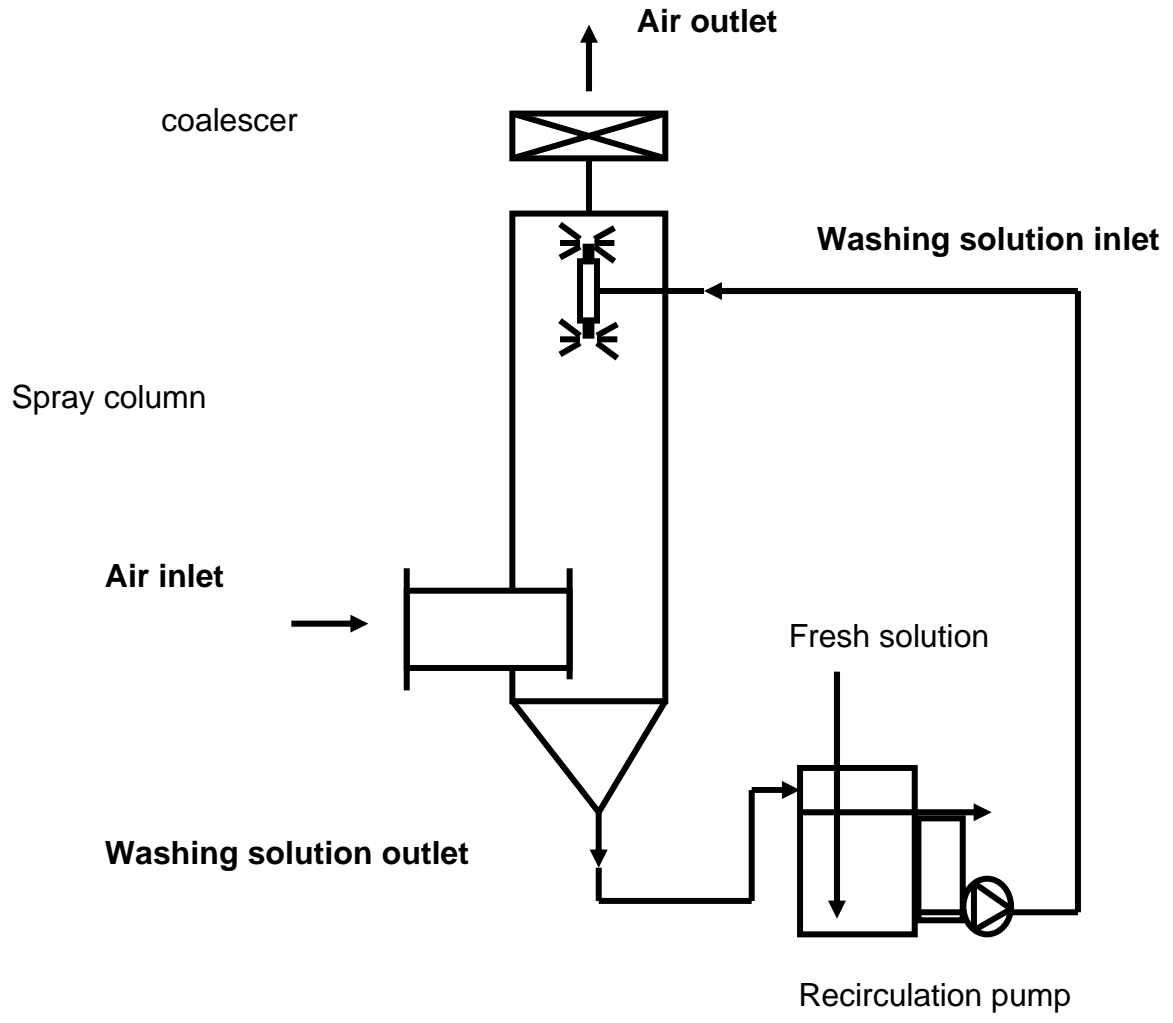


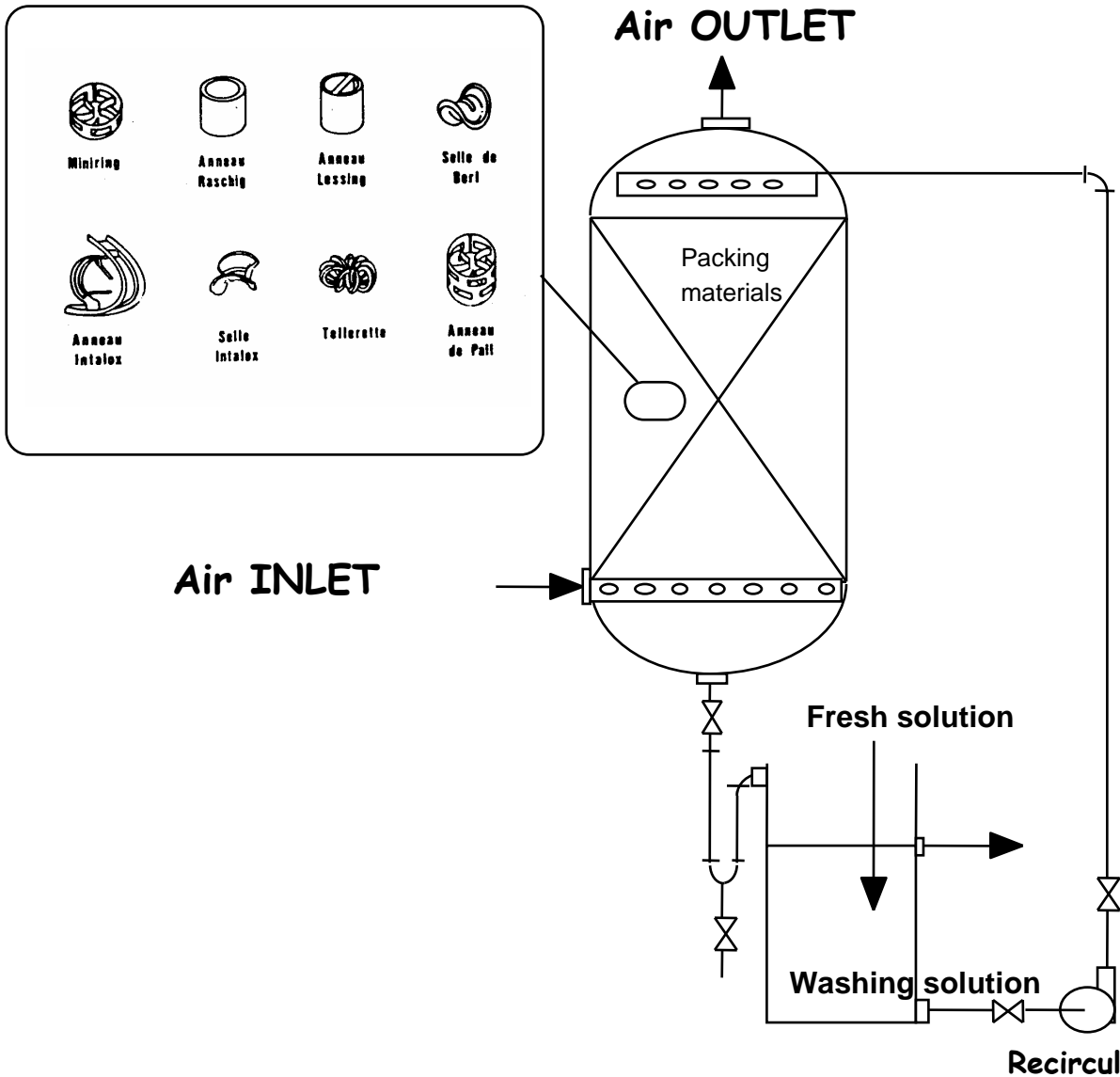
Chemical reaction





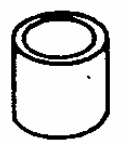
- **Spray column**
- **Packed column**







Miniring



Anneau  
Raschig



Anneau  
Lessing



Selle de  
Bori



Anneau  
Intalox



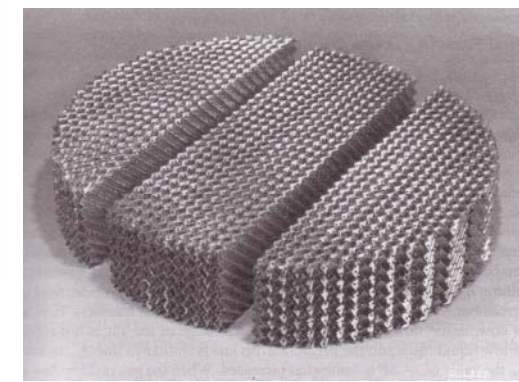
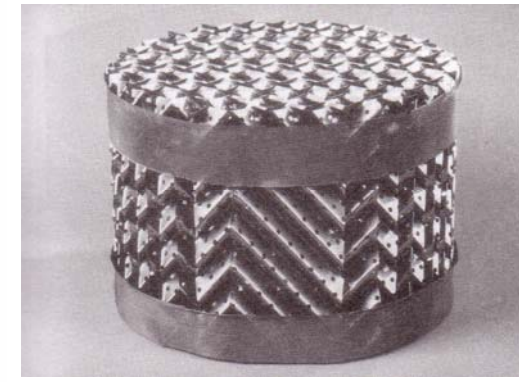
Selle  
Intalox



Tellerette



Anneau  
de Pall



- Acid ( $H_2SO_4$ )

- Base (NaOH)



$CO_2$

- Oxidant (NaOCl,  $O_3$ )

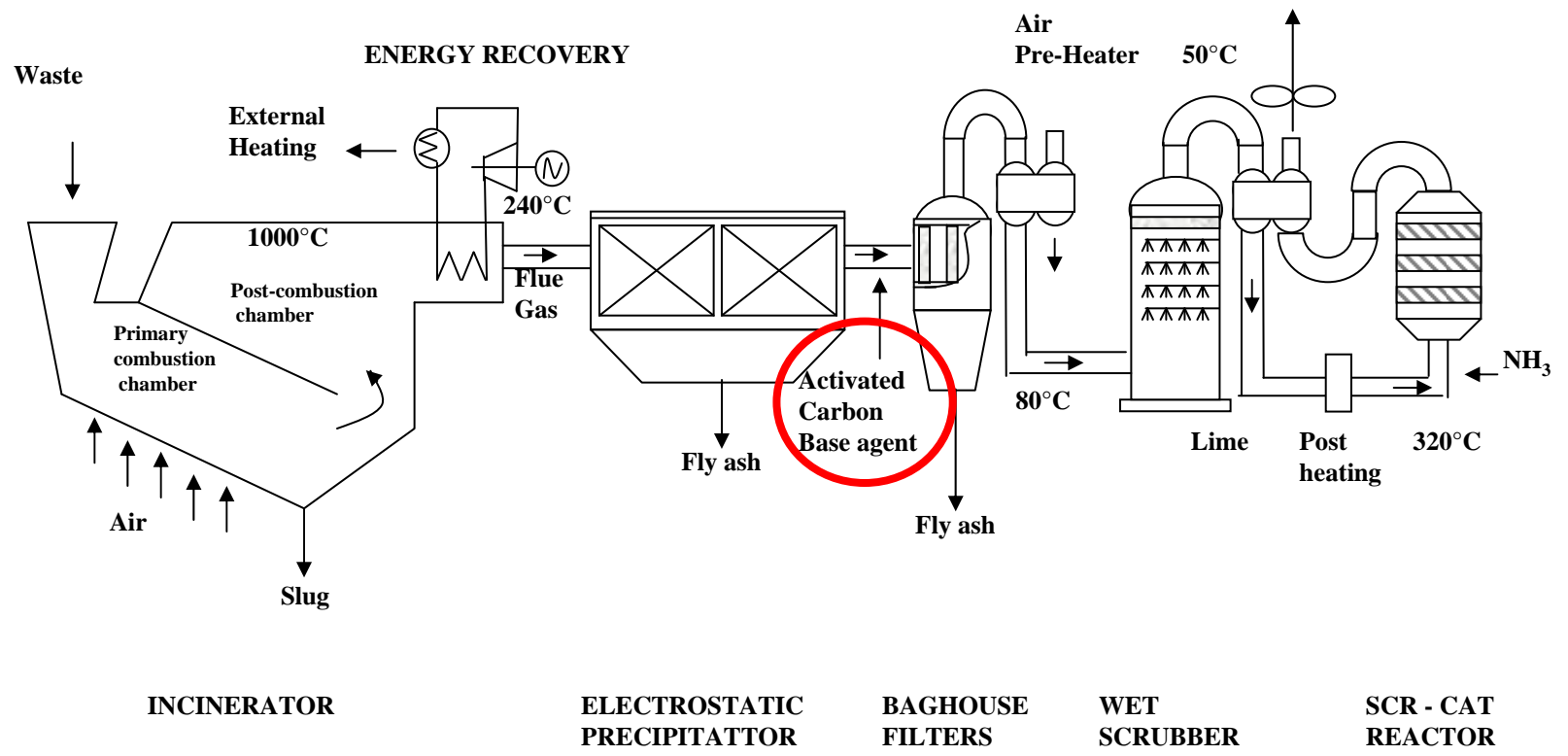
**Liquid solutions treatments**

- **Velocity : 500 - 1000 m/h**
- **Residence time : 0,1 - 10 s**
- **Air temperature : 10 - 80 °C**
- **Concentration : 1 - 50 000 mg/m<sup>3</sup>**
- **Pressure drop : 0,1 - 1 m d 'H<sub>2</sub>O**

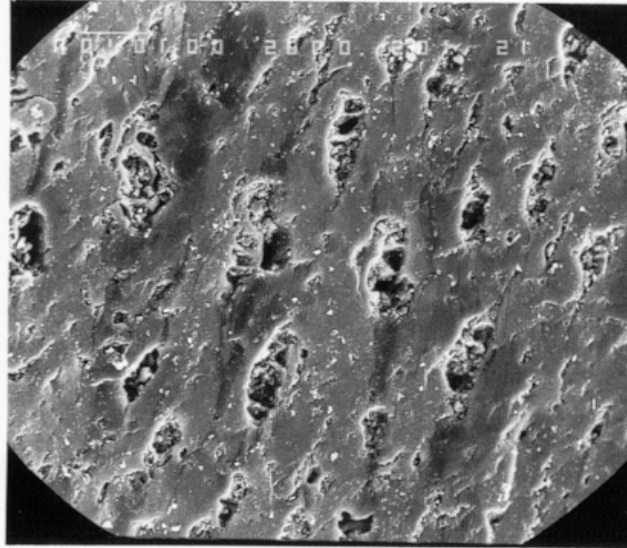
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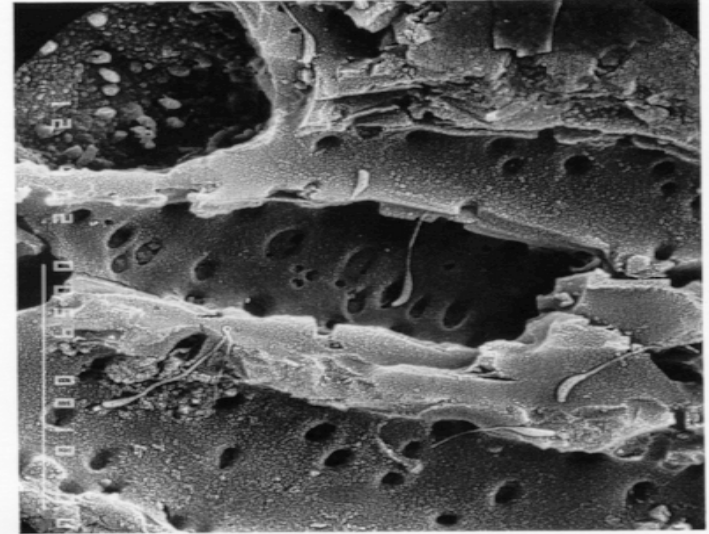




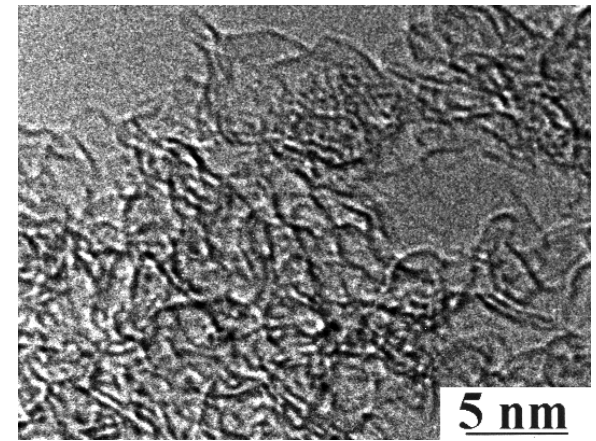
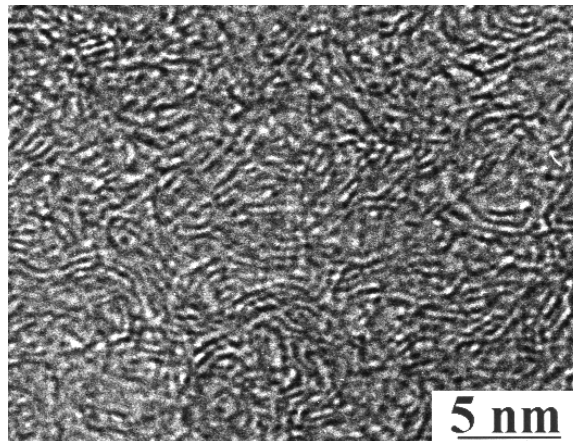
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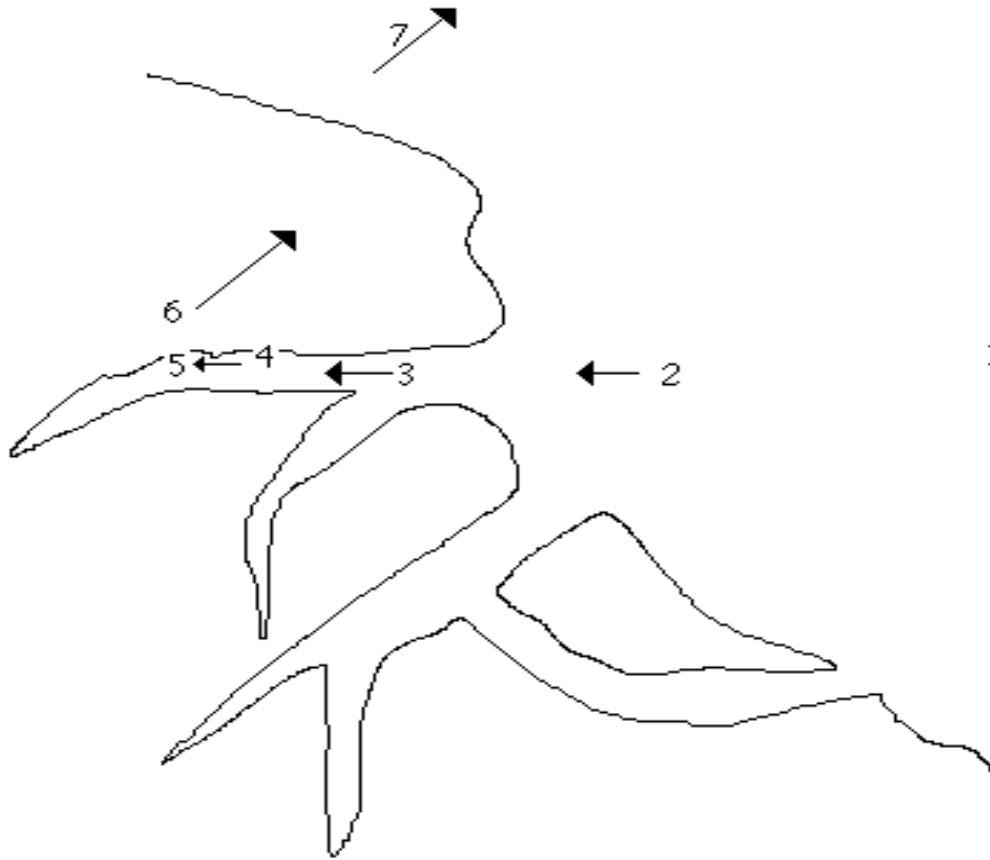
Before activation



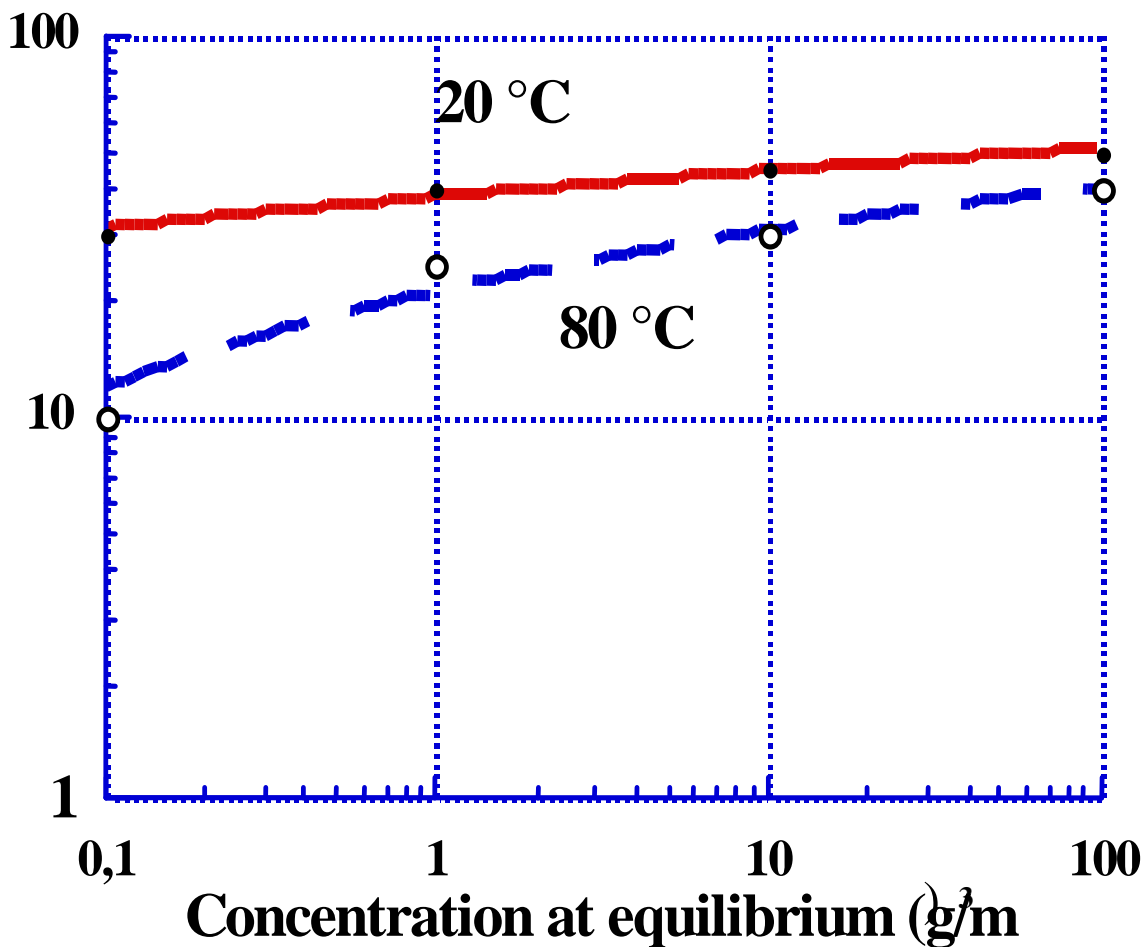
Some strongly activated areas



Pore diameter	nm	dp	
- Macropores			> 50
- Mesopores			2 < d < 50
- Micropores			< 2
Porous volume	cm <sup>3</sup> /g	Vp	0.3 – 0.7
Specific surface area (BET)	m <sup>2</sup> /g	S <sub>BET</sub>	
- non activated			2 - 20
- activated			500 - 2000



Adsorption capacity (%)

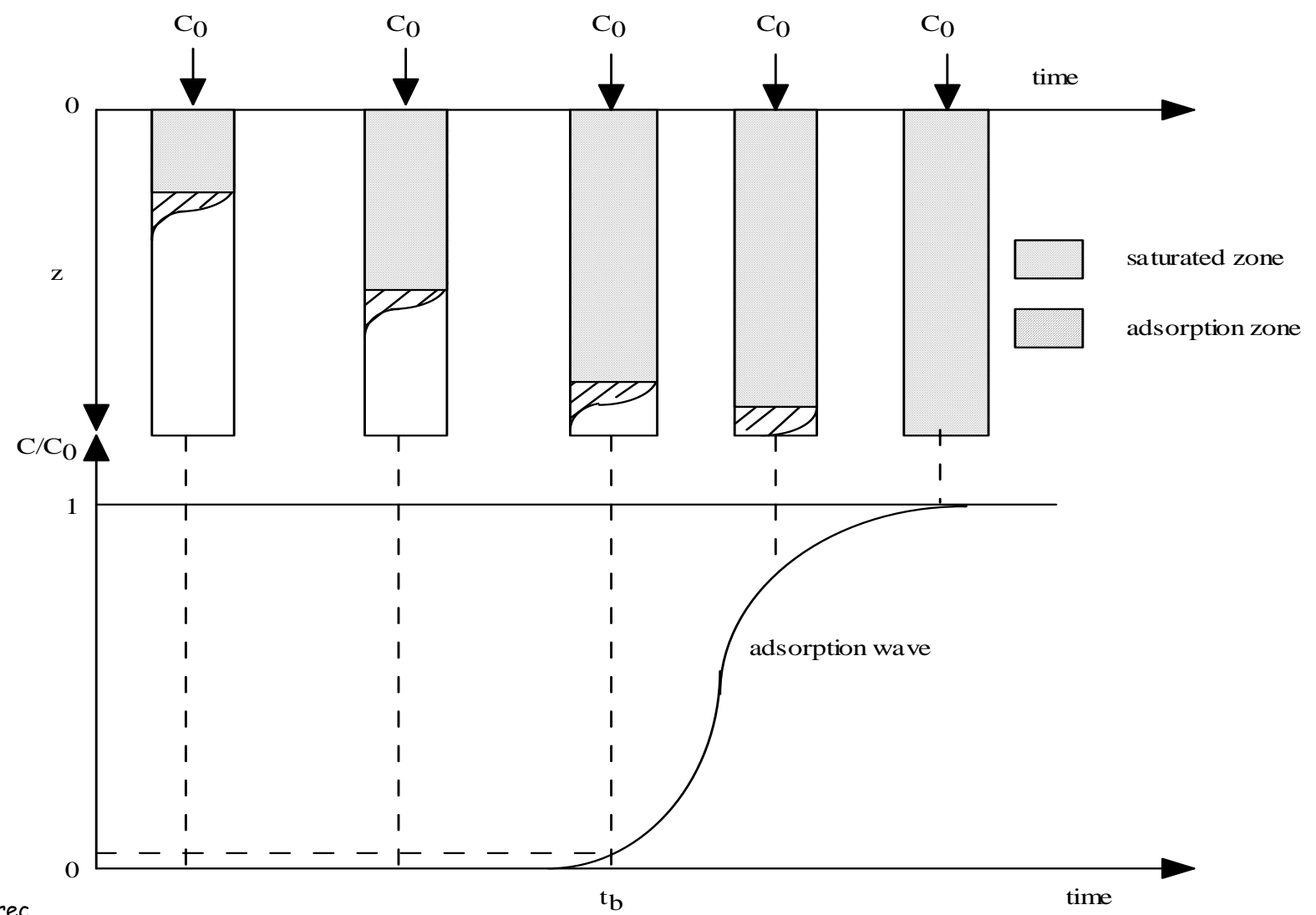


$$q_e = KC_e^{1/n}$$

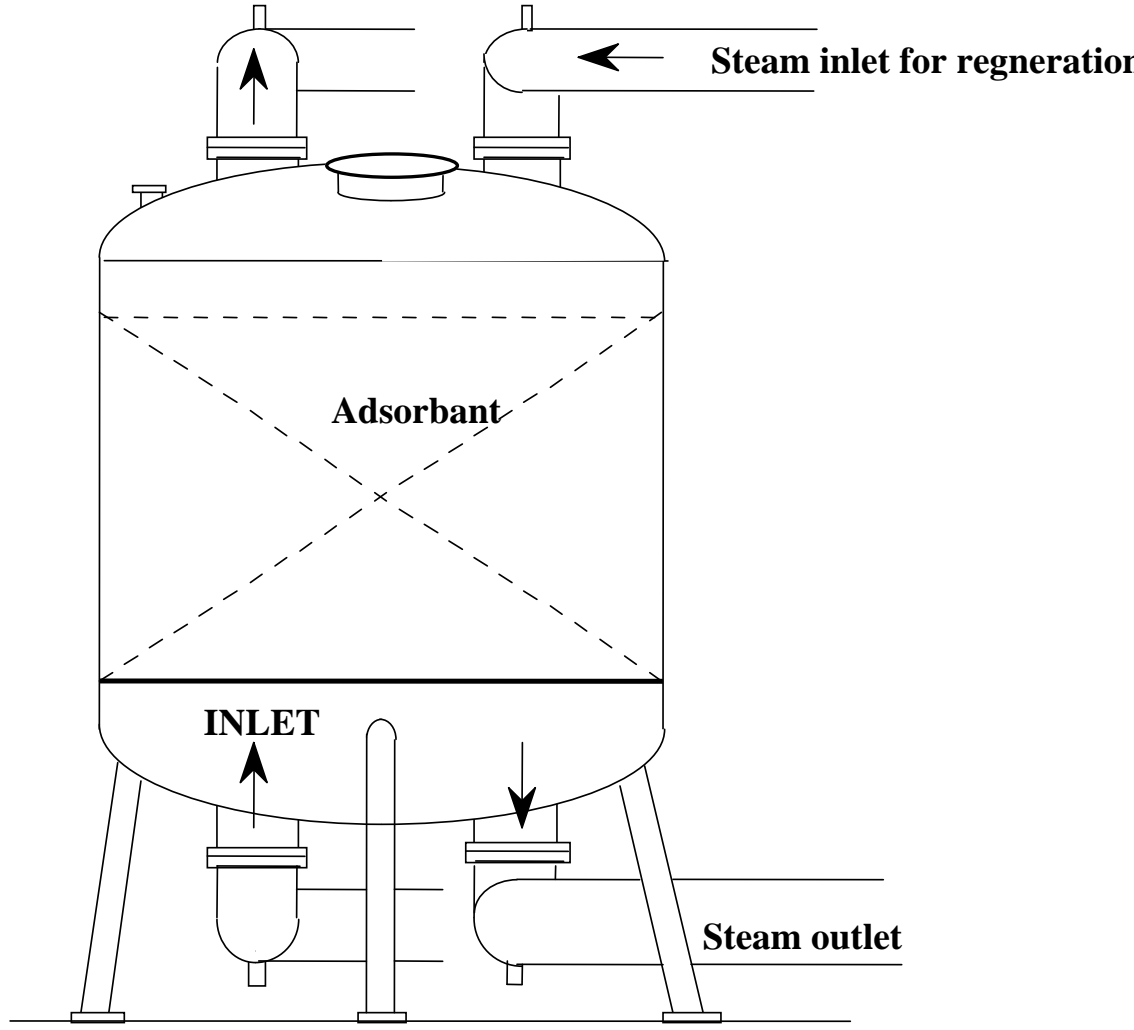
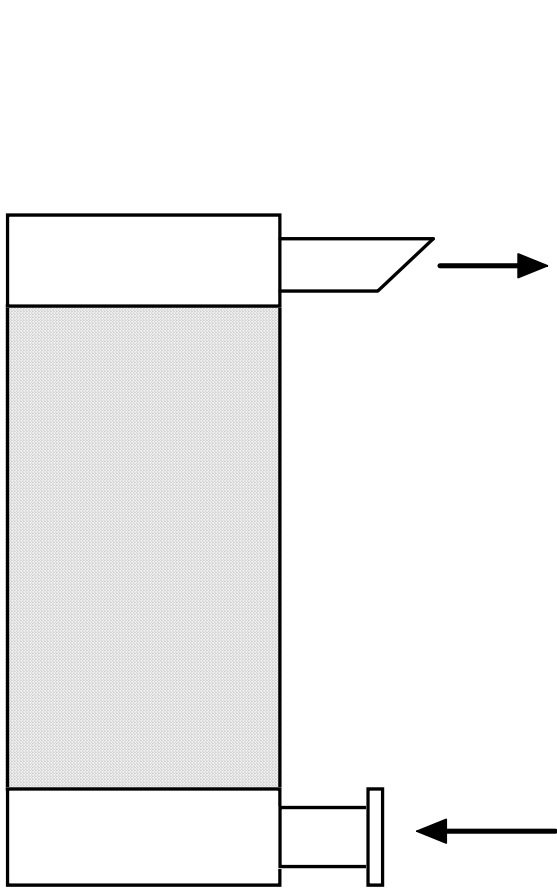
$$q_e = \frac{bq_m C_e}{1 + bC_e}$$

$$q_e = \frac{bq_m C_e^{1/n}}{1 + bC_e^{1/n}}$$

$$q_e = f(C_e)$$



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$$t_b = \frac{N_0}{C_0 U_0} (z - z_0)$$

$$C(t) = \left[ \frac{C_0^{n-1}}{1 + A e^{-rt}} \right]^{1/n-1}$$

## Modeling breakthrough curves Homogeneous surface diffusion model

Purpose	Equation
<b>Solid phase mass balance</b>	$\frac{\partial q}{\partial t} = \frac{D_s}{r^2} \frac{\partial}{\partial r} \left[ r^2 \frac{\partial q}{\partial r} \right]$
<b>Initial condition</b>	$q = 0 \quad (0 \leq r \leq R, t = 0)$
<b>Boundary conditions</b>	$\frac{\partial q}{\partial r} = 0 \quad (r = 0, t \geq 0)$ $\frac{\partial q}{\partial t} = \frac{k_f}{\rho_a D_s \phi} C(t) - C_s(t) \quad (r = R, t)$
<b>Liquid phase mass balance</b>	$V \frac{\partial C}{\partial z} = \frac{\partial C}{\partial t} + \frac{3k_f(1-\epsilon)}{R\phi\epsilon} (C - C_s)$
<b>Initial condition</b>	$C = 0 \quad (0 \leq z \leq H, t < \tau)$
<b>Boundary condition</b>	$C = C_0(t) \quad (z = 0, t \geq 0)$
<b>Freundlich isotherm equation</b>	$q = K [C_s(t)]^{1/n} \quad (r = R, t)$

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